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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/549,585	12/01/2005	Seiko Hirayama	F-8766	1212
JORDAN AND HAMBURG LLP 122 EAST 42ND STREET			EXAMINER	
			JOHNSON, KEVIN M	
SUITE 4000 NEW YORK, NY 10168			ART UNIT	PAPER NUMBER
			1793	
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			02/04/2009	PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

	Application No.	Applicant(s)
	10/549,585	HIRAYAMA ET AL.
Office Action Summary	Examiner	Art Unit
	KEVIN M. JOHNSON	1793
The MAILING DATE of this communication a Period for Reply	ppears on the cover sheet with the	correspondence address
A SHORTENED STATUTORY PERIOD FOR REF WHICHEVER IS LONGER, FROM THE MAILING  - Extensions of time may be available under the provisions of 37 CFR after SIX (6) MONTHS from the mailing date of this communication.  - If NO period for reply is specified above, the maximum statutory perions a finite or period for reply within the set or extended period for reply will, by state Any reply received by the Office later than three months after the main earned patent term adjustment. See 37 CFR 1.704(b).	DATE OF THIS COMMUNICATION 1.136(a). In no event, however, may a reply be tile of will apply and will expire SIX (6) MONTHS from ute, cause the application to become ABANDONE	N. mely filed the mailing date of this communication. ED (35 U.S.C. § 133).
Status		
1) ☐ Responsive to communication(s) filed on 17 2a) ☐ This action is FINAL. 2b) ☐ This action is FINAL. 2b) ☐ This action is application is in condition for allow closed in accordance with the practice under	nis action is non-final. vance except for formal matters, pro	
Disposition of Claims		
4) ☐ Claim(s) 1-13 is/are pending in the application 4a) Of the above claim(s) is/are withdrest is/are allowed. 5) ☐ Claim(s) is/are allowed. 6) ☐ Claim(s) 1-13 is/are rejected. 7) ☐ Claim(s) is/are objected to. 8) ☐ Claim(s) are subject to restriction and compared to the application Papers	rawn from consideration.	
9) The specification is objected to by the Examination The drawing(s) filed on is/are: a) and a specificant may not request that any objection to the Replacement drawing sheet(s) including the correction.	ccepted or b) objected to by the ne drawing(s) be held in abeyance. Se ection is required if the drawing(s) is ob	e 37 CFR 1.85(a). ejected to. See 37 CFR 1.121(d).
11) The oath or declaration is objected to by the □	Examiner. Note the attached Office	e Action or form PTO-152.
Priority under 35 U.S.C. § 119		
12) Acknowledgment is made of a claim for foreign     a) All b) Some * c) None of:     1. Certified copies of the priority docume     2. Certified copies of the priority docume     3. Copies of the certified copies of the prapplication from the International Bure     * See the attached detailed Office action for a lie.	nts have been received. Ints have been received in Applicat Iority documents have been receiveau (PCT Rule 17.2(a)).	ion No ed in this National Stage
Attachment(s)  1) Notice of References Cited (PTO-892)  2) Notice of Draftsperson's Patent Drawing Review (PTO-948)  3) Information Disclosure Statement(s) (PTO/SB/08)  Paper No(s)/Mail Date	4)  Interview Summary Paper No(s)/Mail D 5)  Notice of Informal F 6)  Other:	ate

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## **DETAILED ACTION**

## **Priority**

1. Receipt is acknowledged of papers submitted under 35 U.S.C. 119(a)-(d), which papers have been placed of record in the file. The certified translations of the priority documents have been entered in the record.

2. The previously relied upon Setoguchi reference does not constitute prior art in light of the receipt of the certified translations of the priority documents.

## Claim Rejections - 35 USC § 103

- 3. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
  - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 4. The factual inquiries set forth in *Graham* **v.** *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:
  - 1. Determining the scope and contents of the prior art.
  - 2. Ascertaining the differences between the prior art and the claims at issue.
  - 3. Resolving the level of ordinary skill in the pertinent art.
  - 4. Considering objective evidence present in the application indicating obviousness or nonobviousness.
- 5. This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation

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under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

6. Claims 1, 4-7 and 9 are rejected under 35 U.S.C. 103(a) as being unpatentable over Oshio et al. (US 6096243) in view of Morihito et al. (JP 2000-144129).

In regard to <u>claim 1</u>, Oshio teaches a phosphor of the Ba<sub>0.9</sub>Eu<sub>0.1</sub>MgAl<sub>10</sub>O<sub>17</sub> type (example 1). Oshio fails to teach the inclusion of an element selected from the group including In, W, Nb, Bi, Mo, Ta, Th and Pb.

Morihito teaches a phosphorescent material comprising Eu as an activator, in which the performance of the phosphor is improved by the inclusion of a coactivator that is preferably Nb (abstract, claim 3).

It would have been obvious to one skilled in the art at the time of the invention to utilize a Nb coactivator in the phosphor material taught by Oshio. Such a modification would have been motivated by the teaching in Morihito that phosphor compounds that utilize an Eu activator achieve increased performance with the addition of an Nb coactivator.

In regard to <u>claim 4</u>, Morihito teaches that the amount of Nb coactivator in the phosphor is preferably between 0.0001 and 1 mol-% (paragraph 5).

In regard to <u>claim 5</u>, Oshio teaches that the phosphor may have the formula  $(Ba_aSr_b)Mg_eAl_{10}O_{17}$  where 0 < a < 1,  $0 \le b \le 1$  and  $0 < e \le 1$  (claim 73). The amount of Eu

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activator preferably used is seen in the exemplary embodiments in Oshio,  $Ba_{0.9}Eu_{0.1}MgAI_{10}O_{17}$ .

In regard to <u>claim 6</u>, it would necessarily follow that the phosphor obviated by the teachings of Oshio and Morihito would have a whiteness value of at least 85 as expressed in terms of W value.

In regard to <u>claim 7</u>, Oshio teaches that the phosphor is produced by firing a mixture of precursors in a reducing atmosphere (column 6, lines 43-46).

In regard to <u>claim 9</u>, the mixture of precursors is fired in an oxidizing atmosphere prior to being fired in a reducing atmosphere in the process taught by Oshio (column 6, lines 5-8).

7. Claims 2, 3, 8 and 10-13 are rejected under 35 U.S.C. 103(a) as being unpatentable over Oshio and Morihito as applied to claims 1 and 7 above, and further in view of Bouchard et al. (US 3753759).

In regard to <u>claim 2</u>, Oshio teaches that after the phosphor is produced in the reducing step, it is rinsed and then dried with an oven (column 6, lines 64-67). It is not disclosed that the drying takes place in an oxidizing atmosphere.

Bouchard teaches that it is common to dry phosphors rinsed with deionized water in heated air (column 4, lines 17-21).

It would have been obvious to one skilled in the art at the time of the invention that the drying described in Oshio takes place in an oxidizing environment. This conclusion is supported by the teaching in Bouchard that after rinsing, phosphors are commonly dried by heating in air.

In regard to claims 3, 10 and 12, it is not taught that the coactivator additive taught by Morihito is added to the phosphor taught by Oshio after the mixture has been fired. Oshio teaches that the host raw material mixture is fired in an oxidizing atmosphere to produce the base material for the phosphor and is then combined with the raw material for the Eu activator. The mixture is then fired in a reducing atmosphere, before being washed and dried in an oxidizing atmosphere to produce the phosphor. It would have therefor been obvious to one skilled in the art at the time of the invention to fire the base material and Eu mixture before adding the Nb coactivator precursor to the mixture. Such a modification would have been motivated by the teaching in Oshio that activators are beneficially added to an already fired raw material mixture.

In regard to <u>claim 8</u>, Oshio teaches that after firing in a reducing atmosphere, the phosphor is washed and then dried. It is not disclosed that the drying takes place in an oxidizing atmosphere.

Bouchard teaches that it is common to dry phosphors rinsed with deionized water in heated air (column 4, lines 17-21).

It would have been obvious to one skilled in the art at the time of the invention that the drying described in Oshio takes place in an oxidizing environment. This conclusion is supported by the teaching in Bouchard that after rinsing, phosphors are commonly dried by heating in air.

In regard to <u>claim 11</u>, Morihito teaches that multiple additives selected from Nb, Mo, Ta, W and Bi may be used as coactivators in alkaline earth metal phosphors. It

would therefor be obvious to one skilled in the art at the time of the invention that additional coactivators could be added to a fired product that contains a coactivator, as the in the process taught by Oshio the activator is added to a fired mixture.

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In regard to <u>claim 13</u>, Oshio teaches that to produce the phosphor that contains Eu effectively, the precursor mixture must be fired in a reducing atmosphere so that divalent Eu is present as the activator. It would therefor be obvious to one skilled in the art at the time of the invention that the fired mixture to which the coactivator taught by Morihito is added must have been previously fired in a reducing atmosphere.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to KEVIN M. JOHNSON whose telephone number is (571)270-3584. The examiner can normally be reached on Monday-Friday 7:30 AM to 5:00pm.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Jerry Lorengo can be reached on 571-272-1233. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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/J.A. LORENGO/ Supervisory Patent Examiner, Art Unit 1793 /Kevin M Johnson/ Examiner, Art Unit 1793